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Name. Compound

Formula. 1. Carbon

dioxide. CO 2. 2.

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Carbon monoxide.

*CO. 3. Diphosphorus
pentoxide. P 2 O 5. 4.*

... Covalent

Nomenclature

Worksheet –Answer

Key. Write the correct

name for: 1) As 4 O

10. tetrarsenic

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Answer to: A

compound is found to

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*have 61% carbon,
6.6% hydrogen, and
32.6% oxygen. What is
the empirical formula
of this compound? By
signing up,...* The
Carbon Cycle Quiz –
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Concepts 22.1

Hydrocarbons •

*Because carbon has
four valence electrons,
car-bon atoms always
form four covalent
bonds. • The carbon
atoms in an alkane*

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Guide Answer Key

*can be arranged in a
straight chain or in a
chain that has*

*branches. • Molecules
of hydrocarbons, such
as alkanes, are
nonpolar molecules.*

22.2 Unsaturated

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*Name: ... CCl₄ :
covalent, carbon
tetrachloride . b.*

Fe(OH)₃ : ionic, iron

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(III) hydroxide ...
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Explain your answer.

*Ionic compound's
crystal lattice is more
brittle because it is
composed of charged
ions that must be
aligned to minimize*

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*Study Guide B Section
5: Enzymes*

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*2.3: Carbon-Based
Molecules KEY*

CONCEPT Carbon-

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Carbon-based molecules are

the foundation of life.

Carbon atoms are the

basis of most

molecules that make

up living things. Many

carbon-based

molecules are large

molecules called

polymers that are

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*made of many smaller,
repeating molecules
called monomers.*
There

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Many of them are biologically important. Carbon chemistry is key to life as we know it. The element carbon can exist in three states in nature – as graphite, diamond, or buckey

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balls. The structures

of these forms of

carbon are shown

below: Diamond is

shown on the left. In

diamond, carbon

forms four bonds with

4 other carbons by

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answers 143711. If
you don t see any
interesting for you use
our search form on
bottom. Identify the*

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*following compounds
as ionic compound or
covalent. Name the
following chemical
compounds.*

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Answer to: A

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*have 61% carbon,
6.6% hydrogen, and
32.6% oxygen. What is
the empirical formula
of this compound? By
signing up,...* The
Carbon Cycle Quiz –
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*Name: ... CCl₄ :
covalent, carbon
tetrachloride . b.*

Fe(OH)₃ : ionic, iron

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(III) hydroxide ...
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Explain your answer.

*Ionic compound's
crystal lattice is more
brittle because it is
composed of charged
ions that must be
aligned to minimize*

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*Look at the formula of
the following chemical
compound and answer*

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the questions below:

NaHCO₃ 3. a. How many elements are in this equation? 4. b. List the elements in this equation. sodium, hydrogen, carbon, oxygen. c. How many total atoms are in this equation? 6. 13.

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Balance these

chemical equations ___

2 ___ Cu + O₂ ----->

___ CO₂

**SG 2.3 Carbon-based
molecules - Weebly**

Study Guide B Section

5: Enzymes

REINFORCEMENT

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*2.3: Carbon-Based
Molecules KEY*

CONCEPT Carbon-based molecules are the foundation of life. Carbon atoms are the basis of most molecules that make up living things. Many carbon-based

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molecules are large

molecules called

polymers that are

made of many smaller,

repeating molecules

called monomers.

There

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Answer Key SECTION
1. ATOMS, IONS,
AND MOLECULES 1.
An element is a*

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certain type of atom.

2. Sketch should resemble one of the illustrations in Figure 1.2 in Section 1.

Nucleus should be in the center with protons and neutrons. Electrons should be labeled in a ring

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*surrounding the
nucleus. 3.*

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Organic ... The
structure of the
compound contains 6-

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*membered carbon
chain and functional
groups is alcohol at
C3. So, the parent
chain is named as 3-
hexanol. Two ...*

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Key 1. Write the
formula for the
following binary ionic
compounds. a. CaCl_2
b. NaF c. MgO d. AlF_3
e. Al_2S_3 f. Li_2O g.
 PbO_2 h. CuF_2 i. Na_3P
j. Li_3N_2 . Name*

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the following binary

ionic compounds. a.

strontium iodide b.

lithium oxide c. silver

fluoride d. iron(III)

chloride e. copper(II)

oxide 3.

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