

Ebooks Is The  
Solution Unsaturated  
Saturated Or  
Supersaturated  $\text{KNO}_3$

**Is The  
Solution  
Unsaturated  
Saturated**

Ebooks Is The  
Solution Unsaturated  
Saturated Or  
**Or** Supersaturated  $\text{Kno}_3$

**Supersaturated  
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**Solution**  
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K<sub>NO</sub><sub>3</sub> Book [PDF]  
**Solutions Brain**  
**Genie Flashcards |**

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**Quizlet**  
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*The concentration of solute in an unsaturated solution is higher than that in a saturated solution at the same temperature.*  
*False. ... A supersaturated solution contains 100*

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Saturated Or  
*g of KNO<sub>3</sub>...*  
Supersaturated Kno<sub>3</sub>

## **Solubility and Solubility Curves - Video & Lesson ...**

*A saturated solution is  
a solution that  
contains the maximum  
amount of dissolved  
solute. If a solution*

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Saturated Or  
*does not contain the  
Supersaturated KNO<sub>3</sub>  
maximum amount of  
solute, it is  
unsaturated .*

**can you explain  
every question ? I  
did the ones i |**

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*Transcribed image*

*text: 150 140 130 120*

*110 100 NONO KNO*

*6. Use the solubility  
curve to answer the  
following (screencast  
this onet) Look at the  
position of the star. It  
is describing a 100 mL*

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Solution Unsaturated  
Saturated Or  
*solution of potassium  
nitrate a) This solution  
is (highlight one)  
saturated,  
unsaturated,  
supersaturated.*

**NCERT Exemplar  
Class 12 NCERT  
Exemplar Chemistry**



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**Solutions ...**  
Supersaturated  $\text{KNO}_3$

4. A beaker contains a solution of a substance 'A'. Precipitation of substance 'A' takes place when a small amount of 'A' is added to the solution. The solution is \_\_\_\_\_. (i) saturated (ii)

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Saturated Or  
*supersaturated (iii)*  
*unsaturated (iv)*  
*concentrated.*

*Solution: Option (ii) is  
the answer. 5.*

**Chemistry Enhanced**

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Saturated Or  
**Scope & Sequence**  
Supersaturated  $\text{KNO}_3$

*5. By cooling the saturated solution rapidly and carefully, you have created a supersaturated solution. The cooled solution actually contains more solid than it should contain*

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Solution Unsaturated  
Saturated Or

*at that temperature.*

*Supersaturated  $KNO_3$   
This is a very unstable  
solution, and it will  
not stay*

*supersaturated very  
long. It can be easily  
changed by dropping  
in a seed crystal of the*

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Solution Unsaturated  
Saturated Or  
*solid. 6.*  
Supersaturated  $\text{Kno}_3$

Once more Is The Solution  
Unsaturated Saturated Or  
Supersaturated  $\text{Kno}_3$ , what  
kind of person are you If  
you are essentially one of  
the people behind right of  
entry minded, you will have  
this cd as your reference.

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